

# Supporting Information

## Thiocyanate Anchors for Salt-like Iron(II) Complexes on Au(111): Promises and Caveats

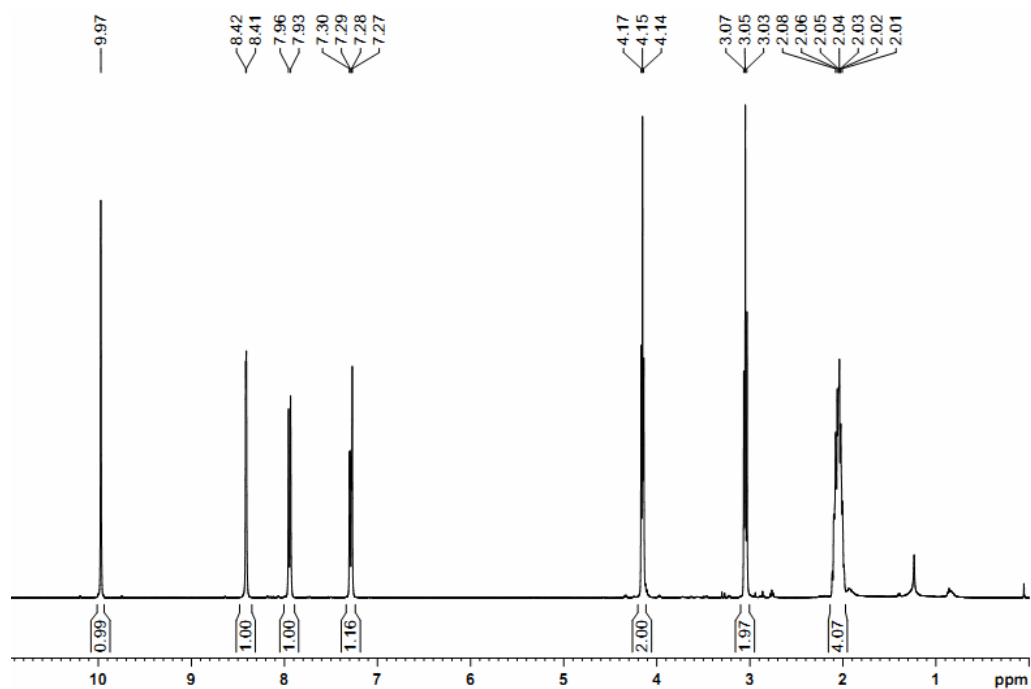
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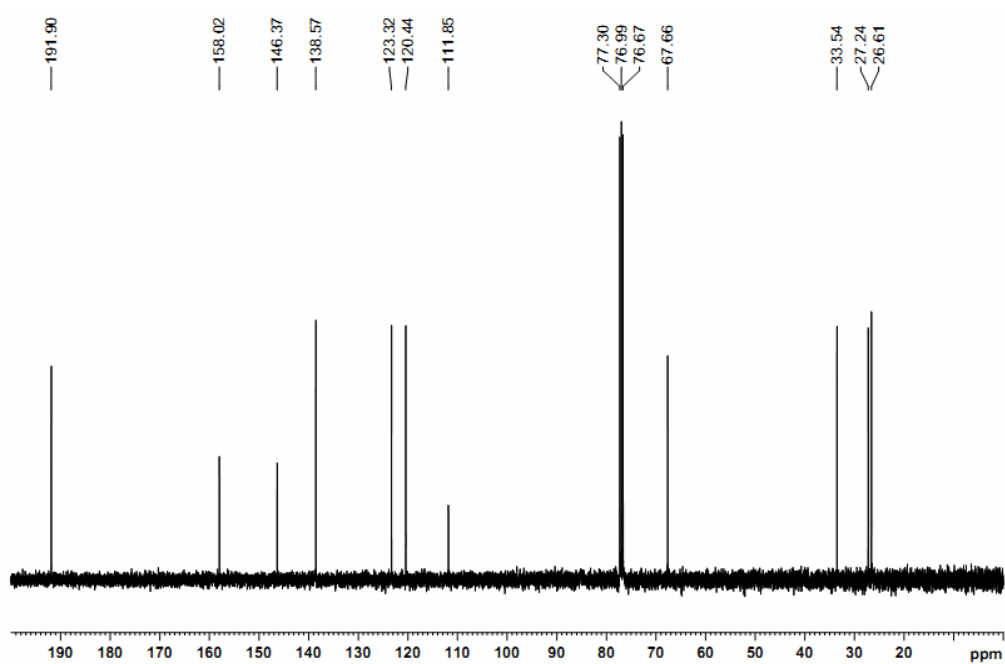
<sup>b</sup> Max-Planck-Institut für Eisenforschung GmbH, Department of Interface Chemistry and Surface Engineering, Max-Planck-Straße 1, 40237 Düsseldorf, Germany.

<sup>c</sup> EaStCHEM School of Chemistry, University of St Andrews, St Andrews KY16 9ST, United Kingdom.

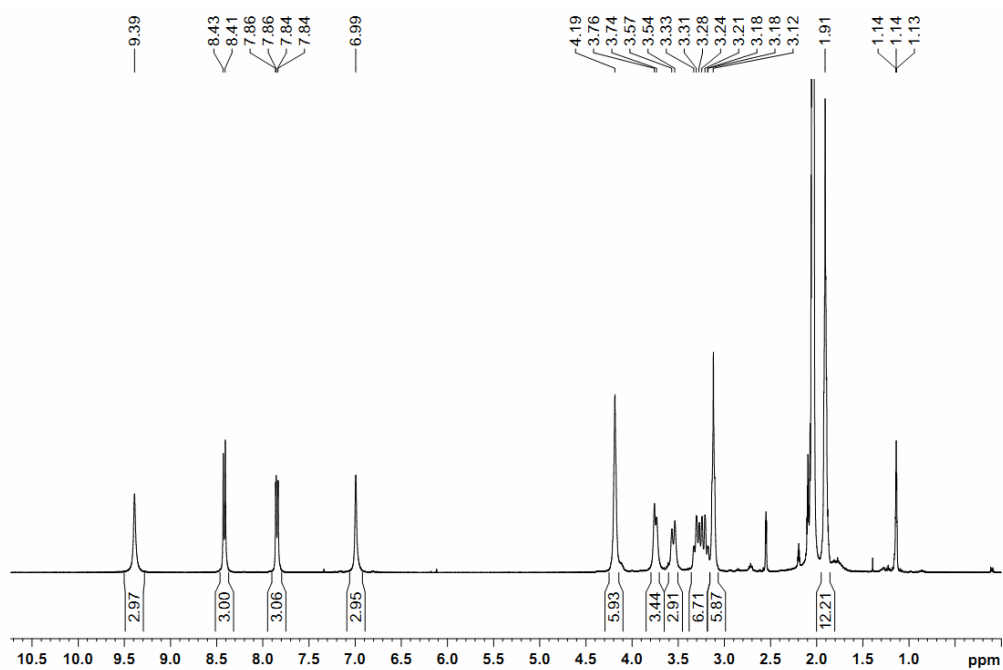
<sup>d</sup> Sasol Technology (UK) Ltd, Purdie Building, North Haugh, St Andrews, KY16 9ST, United Kingdom



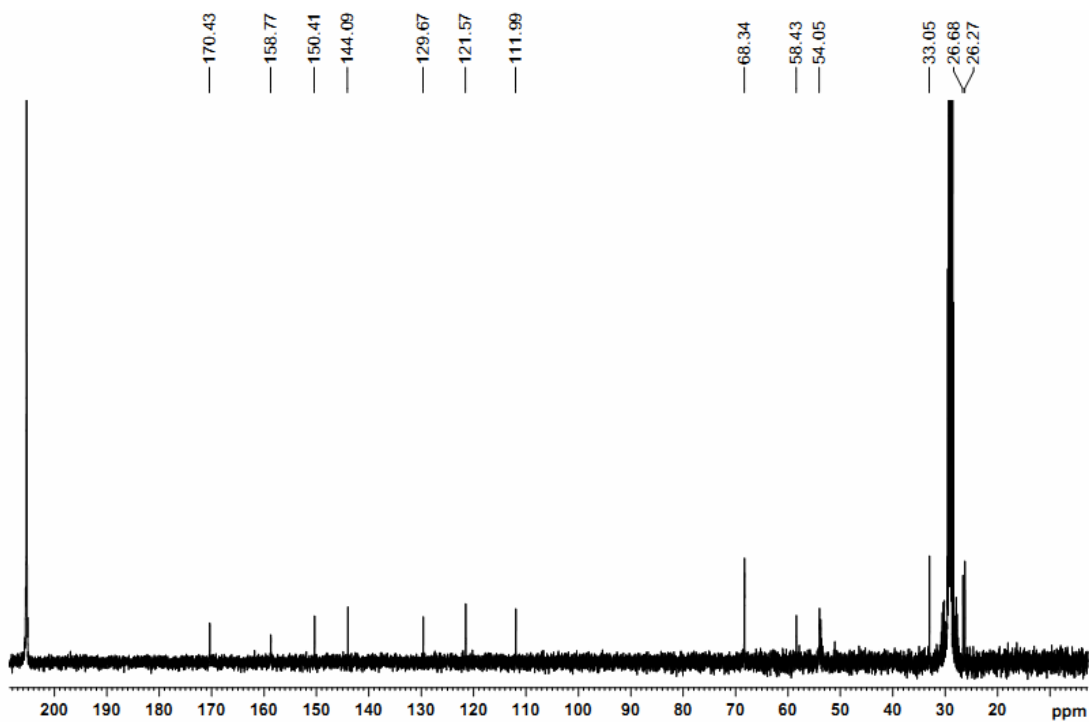
**Figure S1.**  $^1\text{H}$  NMR of **3** in  $\text{CDCl}_3$ .



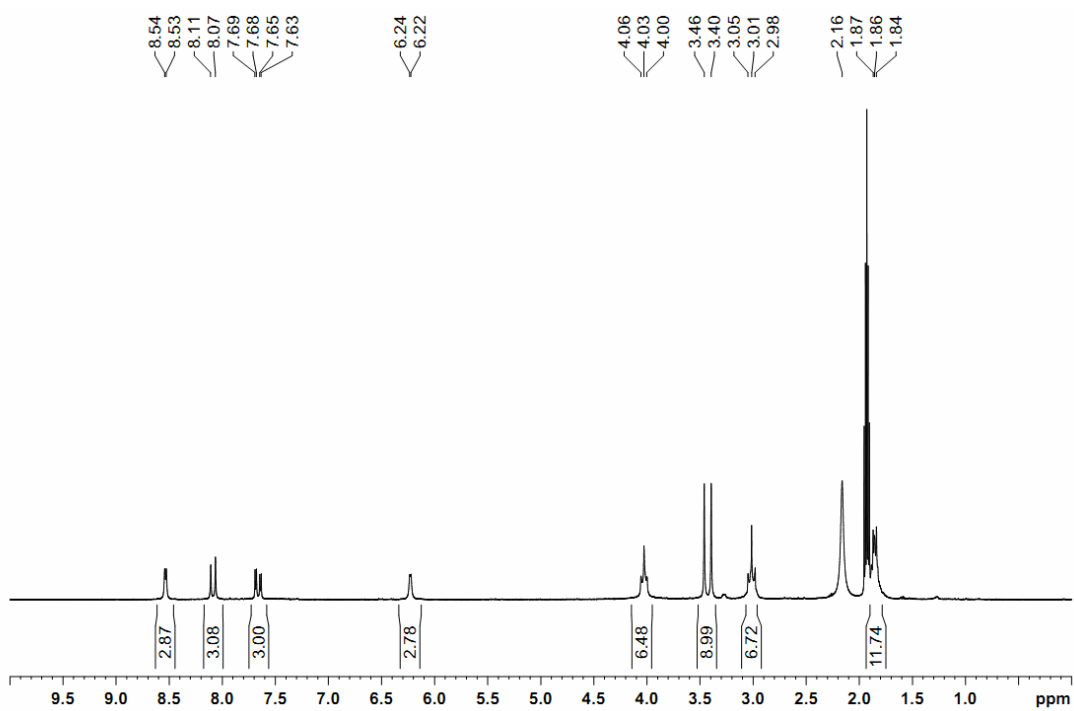
**Figure S2.**  $\{^1\text{H}\}^{13}\text{C}$  NMR of **3** in  $\text{CDCl}_3$ .



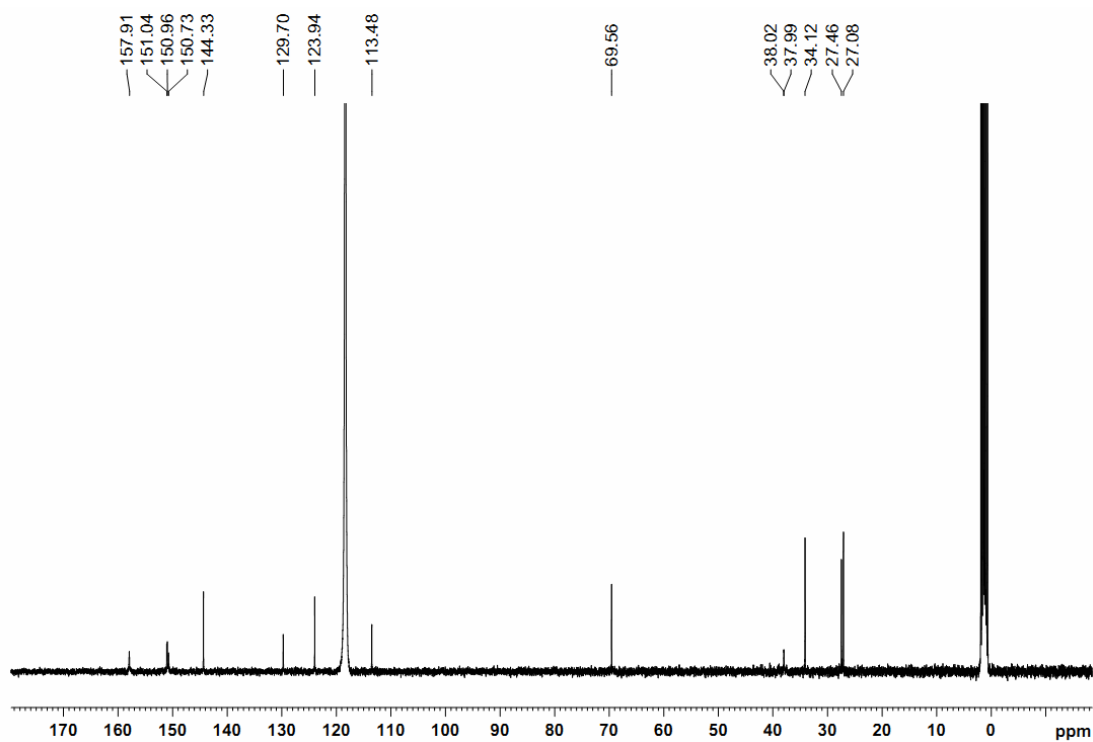
**Figure S3.**  $^1\text{H}$  NMR of **6** in  $[\text{D}_6]\text{acetone}$ .



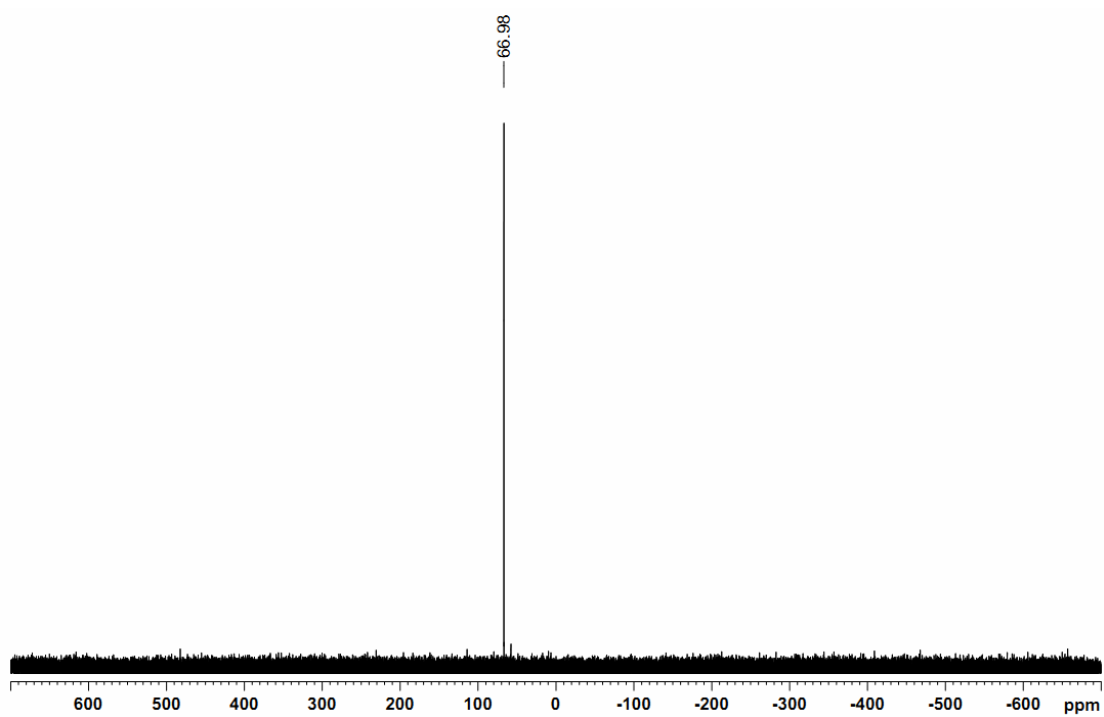
**Figure S4.**  $\{^1\text{H}\}^{13}\text{C}$  NMR of **6** in  $[\text{D}_6]\text{acetone}$ .



**Figure S5.**  $^1\text{H}$  NMR of **7** in  $[\text{D}_3]\text{MeCN}$ .



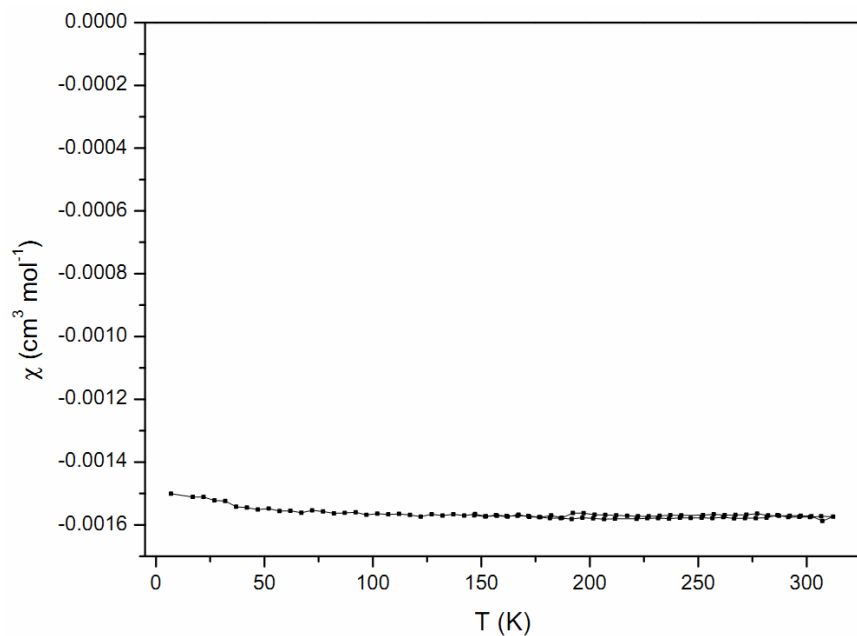
**Figure S6.**  $\{^1\text{H}\}^{13}\text{C}$  NMR of **7** in  $[\text{D}_3]\text{MeCN}$ .



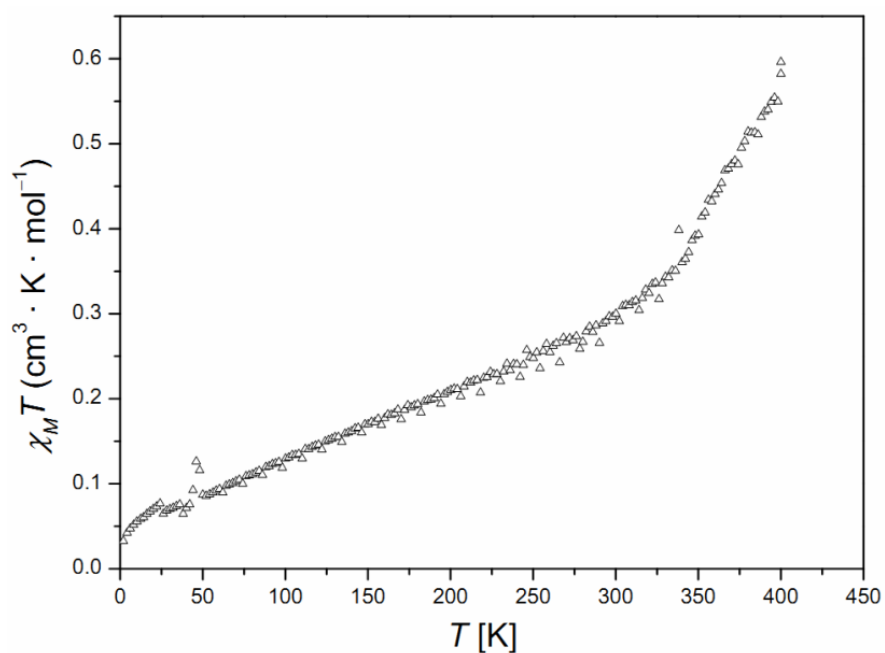
**Figure S7.**  ${}^1\text{H}\text{-}{}^{31}\text{P}$  NMR of **7** in  $[\text{D}_3]\text{MeCN}$ .

## Bulk magnetic behaviour

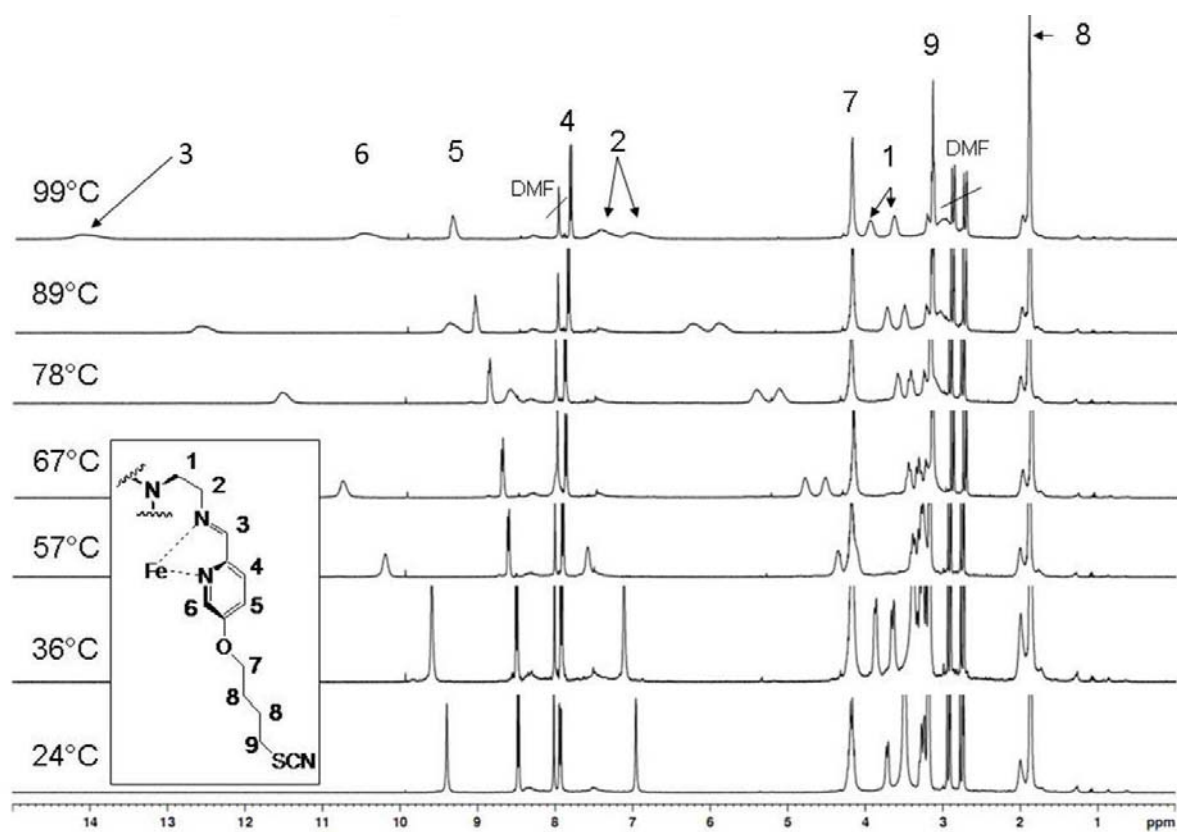
Solid-state variable-temperature magnetic susceptibility measurements were performed using a Quantum Design MPMS-XL5 SQUID (superconducting quantum-interference device) magnetometer operating at 0.1 T. Diamagnetic corrections for the sample and the sample holder were applied.



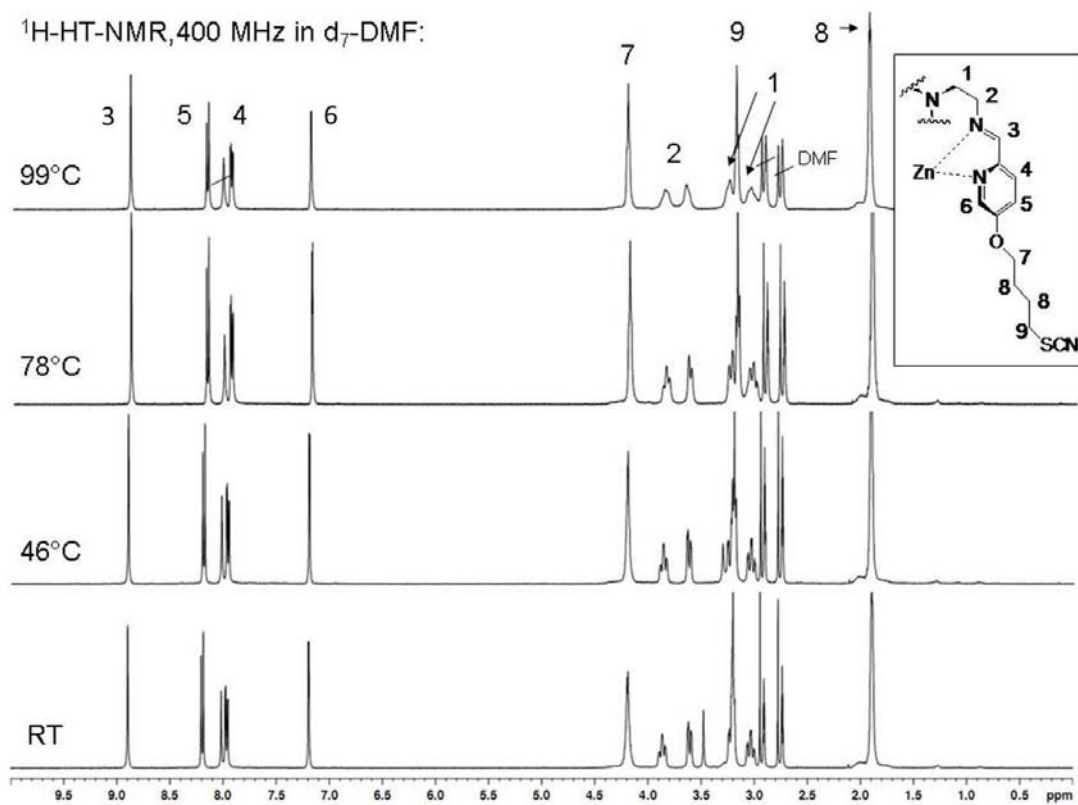
**Figure S8.** SQUID measurement of compound  $6^{\text{Fe}}$ .



**Figure S9.** Magnetic susceptibility  $\chi_M T$  of compound **7**



**Figure S10.** Temperature-dependent  $^1\text{H}$  NMR spectrum of  $6^{\text{Fe}}$  (400 MHz, (D<sub>7</sub>)DMF).



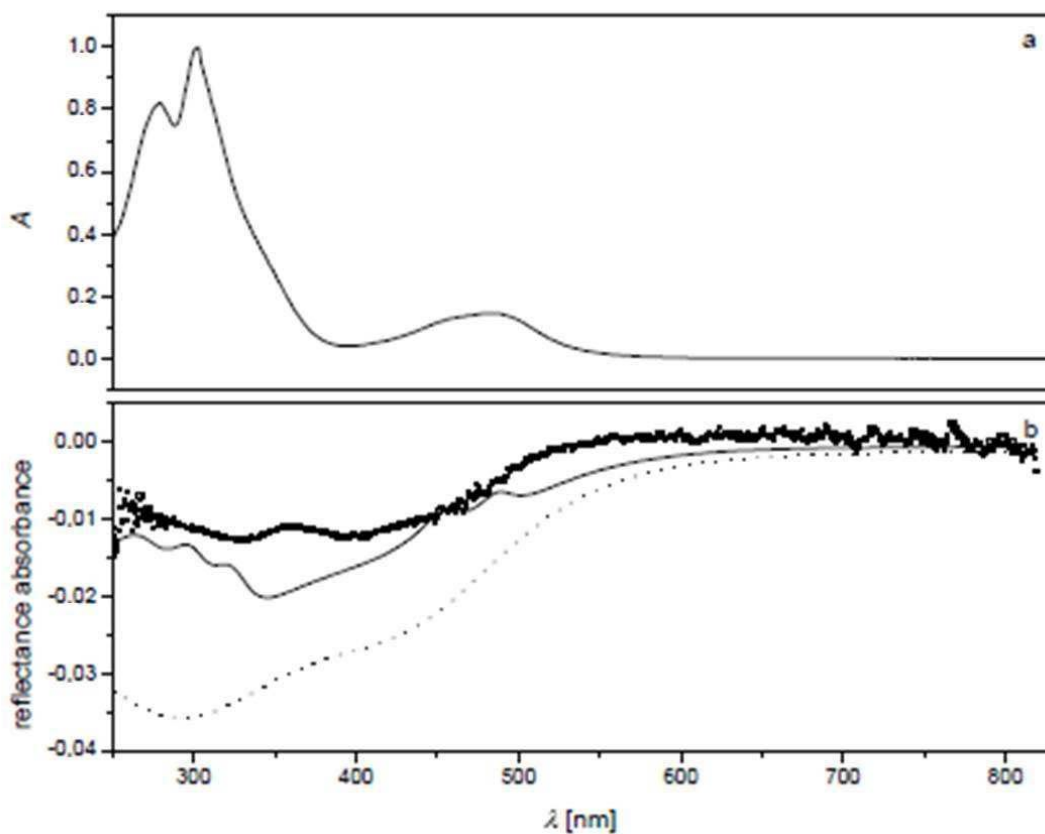
**Figure S11.** Temperature-dependent  $^1\text{H}$  NMR spectrum of the zinc complex,  $6^{\text{Zn}}$  (400 MHz, (D<sub>7</sub>)DMF).

## UV/Vis Reflectance Spectroscopy: Methods and additional spectra

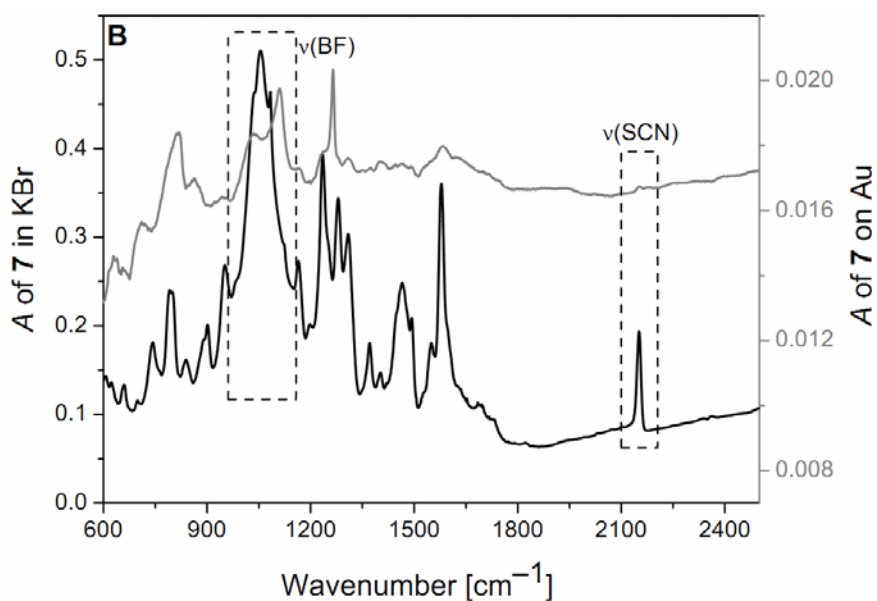
UV/Vis reflectance spectra at an angle of incidence of  $80^\circ$  in p-polarization were obtained using the following procedure: The retarder was removed from the beam path of the spectroscopic ellipsometer, and both polarizer and analyser were set to p-polarization. The raw intensities  $I_{\text{mod}}$  of the modified samples were recorded. Using the same setting for the integration time, the detector dark current  $I_{\text{dark}}$  was measured. In a reference measurement, the reflected intensity  $I_{\text{ref}}$  from an unmodified, freshly prepared gold surface was recorded. The reflectance absorbance was then computed as  $-\log_{10}((I_{\text{mod}} - I_{\text{dark}})/(I_{\text{ref}} - I_{\text{dark}}))$ . P-polarization at high incidence angles was used in order to probe transition dipole moment components that are perpendicular to the surface. Only these can be excited on a metallic surface. Spectroscopic ellipsometry at angles of incidence of  $50^\circ$  and  $70^\circ$  was used to determine the thickness  $d$  of the adsorbed layers. Briefly, the experiment measures the ratio  $r_p/r_s = \tan(\Psi) e^{i\Delta}$  (with  $i = \sqrt{-1}$ ) between amplitude reflection coefficients  $r_p$  for p- and  $r_s$  for s-polarized light. This ratio is expressed using the two ellipsometric angles  $\Psi$  and  $\Delta$ .<sup>S1, S2</sup> The difference between treated samples and an unmodified gold surface was analysed. For a layer which is thin compared to the wavelength of light, with refractive index  $n_s$ , the difference in  $r_p/r_s$  between covered and uncovered surface is directly related to the first order perturbation parameter,  $J_1 = d (1 - n_s^2)(n_s^2 - n_{\text{Au}}^2) / n_s^2$ .<sup>S3</sup> The wavelength-dependent values of the refractive index  $n_{\text{Au}}$  of gold were taken from the literature.<sup>S4</sup>

For the surface modification, an isotropic layer of thickness 2.4 nm and a wavelength-independent refractive index of 1.5 were assumed to be adsorbed to gold. These parameters described the obtained ellipsometric data quite well (see section on ellipsometry in the manuscript). For such a system, the dotted line in Figure 2b shows the same slope in the baseline as present in the experimental data, indicating that this feature is not related to absorptions from a layer adsorbed to the surface. In order to understand the effects that a surface layer of finite absorbance has on the shape of the spectra, simulations were carried out in which the dielectric function of the layer was represented by a sum of five harmonic oscillators. Input parameters for oscillator frequency and width were obtained by fitting the UV/VIS spectrum in solution (Figure 3a and Fig. S13a) with a sum of Lorentzian peaks. Absorption features from the molecules deposited on the surface show as additional peaks on top of the baseline.



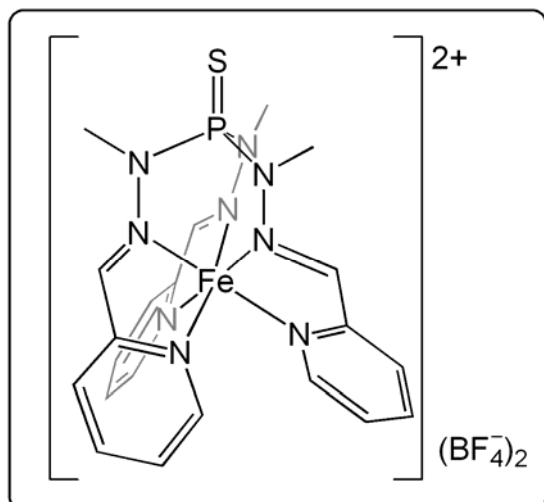


**Figure S12.** UV/VIS spectra of **7**; (a) measured in transmission in methanol solution; (b) measured in reflection on Au(111); dotted line: simulation of a non-absorbing layer (2.4 nm; refractive index 1.5) on Au; black solid line: simulation of a spectrum with 5 harmonic oscillators with resonance frequencies as obtained from (a).

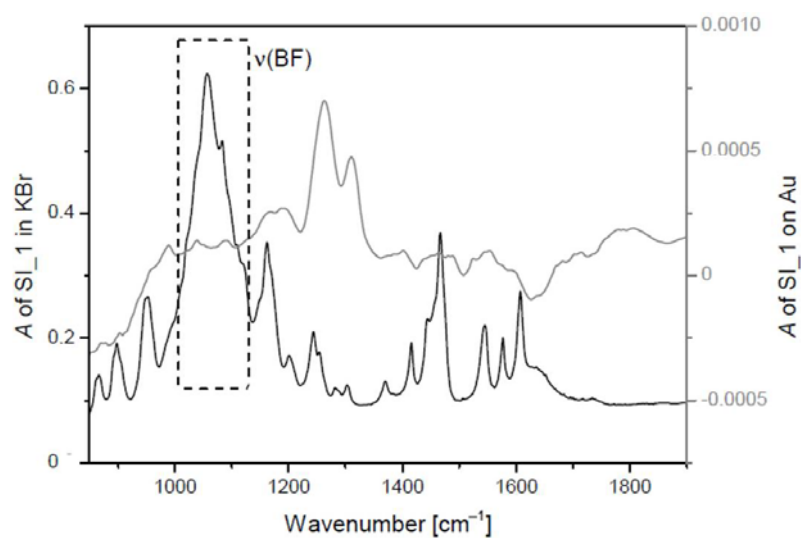


**Figure S13.** IR spectra of compound **7**. Characteristic absorptions are highlighted by dashed boxes. Grey: sample layer on gold. Black: sample measured as KBr disc.

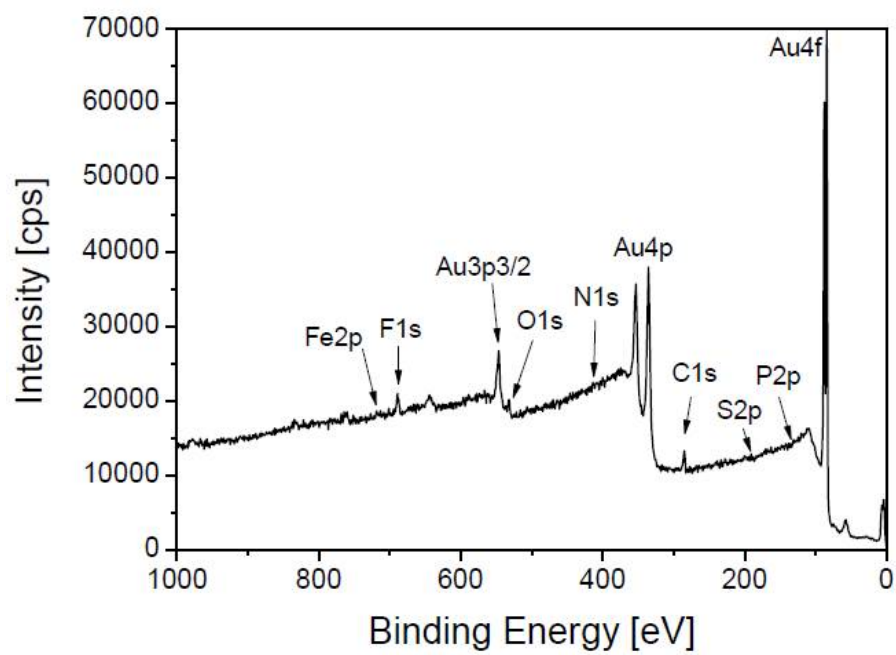
### Reference compound for 7 in control adsorption studies



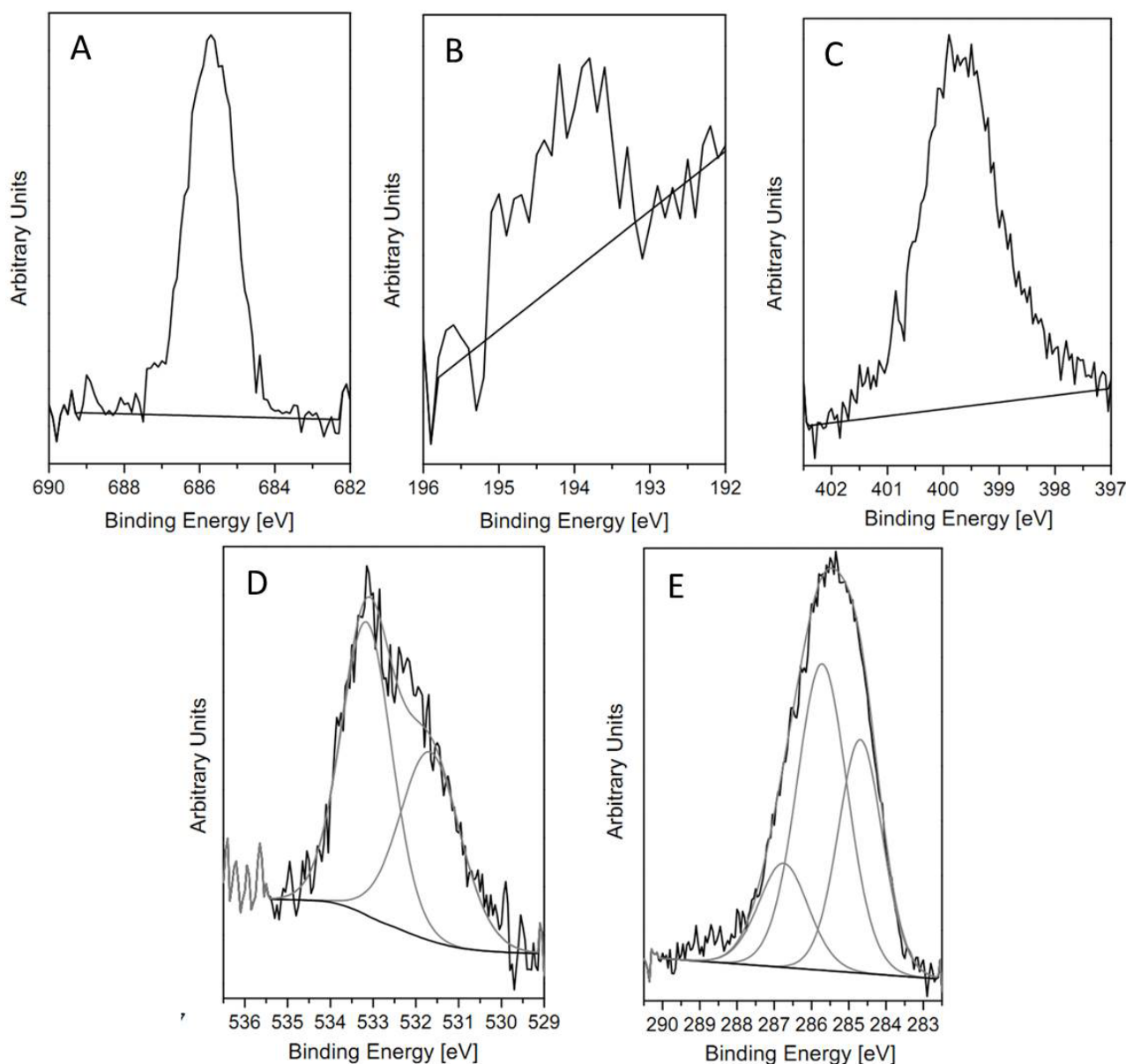
**Figure S14.** Molecular structure of  $[\text{Fe}\{\text{phos}(\text{py})_3\}](\text{BF}_4)_2$  (SI\_1)



**Figure S15.** IR spectra of compound SI\_1. Characteristic absorptions are highlighted by dashed box. Grey: sample layer on gold. Black: sample measured as KBr disc.



**Figure S16.** XPS spectrum (survey scan) of SI\_1 on gold.



**Figure S 17.** XPS spectra of a layer of  $6^{\text{Fe}}$  adsorbed on Au(111): (A) F 1s, (B) B 1s, (C) N 1s, (D) O 1s, (E) C 1s,. All spectra show the baselines used to calculate the intensity of the signals.

## References

- (S1) H. Fujiwara, Principles of Spectroscopic Ellipsometry in *Spectroscopic Ellipsometry: Principles and Applications*, John Wiley & Sons, Chichester, **2007**.
- (S2) R. M. A. Azzam, N. M. Bashara, *Ellipsometry and Polarized Light*, Elsevier Science, Amsterdam, **1999**.
- (S3) J. Lekner, *Theory of Reflection of Electromagnetic and Particle Waves*, M. Nijhoff Publishers, Dordrecht, **1987**.
- (S4) P. G. Etchegoin, E. C. Le Ru, M. Meyer, An analytic model for the optical properties of gold, *J. Chem. Phys.* **2006**, *125*, 164705–164708.